

# Technical Bulletin

## *NO. 747*

**TOPIC: VOLT-HOURS x AMP-HOURS IS NOT EQUAL TO  
WATT HOURS.**

File TEC747b

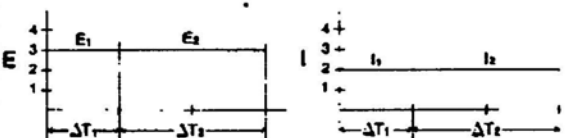
In DynAmp's experience of many years in the electro-chemical industry, there is still one fundamental error concerning Watt-hours calculations. There is not a clear understanding by everyone that if you take Volt-hours x amp-hours, this does not equal Watt-hours. This bulletin attempts to explain why this is true.

$$\left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right) \quad \text{May not be always equal to dc power}$$

Question:  $\left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right)$  equals average dc power, TRUE or FALSE?

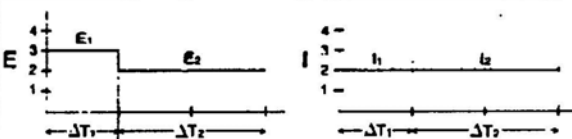
The answer to that question is:

**TRUE when voltage and current are both constant**



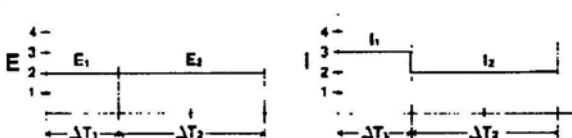
$P_{d-c(aver.)} = \left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right) = 18 \text{ units}$   
 OR  
 $P = E_1 I_1 T_1 + E_2 I_2 T_2 = 18 \text{ units}$

**TRUE when voltage is variable but current is constant**



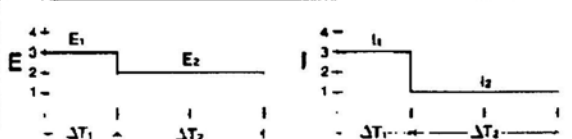
$P_{d-c(aver.)} = \left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right) = 14 \text{ units}$   
 OR  
 $P = E_1 I_1 T_1 + E_2 I_2 T_2 = 14 \text{ units}$

**TRUE when current is variable but voltage is constant**



$P_{d-c(aver.)} = \left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right) = 14 \text{ units}$   
 OR  
 $P = E_1 I_1 T_1 + E_2 I_2 T_2 = 14 \text{ units}$

**FALSE when voltage and current are both variable**



$P_{d-c(aver.)} = \left(\frac{1}{T} \int E dt\right) \cdot \left(\frac{1}{T} \int I dt\right) = 11.6 \text{ units}$   
 but does not correspond to  
 $P = E_1 I_1 T_1 + E_2 I_2 T_2 = 13 \text{ units}$